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KEY TOPICS: Units of Measurement for Physical and Chemical Change; Atoms and Elements; Molecules, Compounds, and Nomenclature; Chemical Reactions and Stoichiometry; Gases; Thermochemistry; The Quantum-Mechanical Model of the Atom; Periodic Properties of the Elements; Chemical Bonding I: Lewis Theory; Chemical Bonding II: Molecular Shapes, Valence Bond Theory, and Molecular Orbital Theory; Liquids, Solids, and Intermolecular Forces; Solutions; Chemical Kinetics; Chemical Equilibrium; Acids and Bases; Aqueous Ionic Equilibrium; Gibbs Energy and Thermodynamics; Electrochemistry; Radioactivity and Nuclear Chemistry; Organic Chemistry I: Structures; Organic Chemistry II: Reactions; Biochemistry; Chemistry of the Nonmetals; Metals and Metallurgy; Transition Metals and Coordination Compounds MARKET: Appropriate for General Chemistry (2 - Semester) courses. Inorganic Chemistry This new textbook brings a fresh and exciting approach to teaching modern inorganic chemistry. 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And with a more traditional and logical organisation of the Organic Chemistry content, this comprehensive text is the source of all the information and practice problems students are likely to need for conceptual understanding, development of problem solving skills, reference and test preparation. Practising Biology A Student Workbook for Biology, 6th Ed. Jean Table of contents continued -- How are water and good transported in plants? -- What do you need to consider in order to grow plants in space (or anywhere else for that matter)? -- How can plant reproduction be modified using biotechnology? -- How do gravity and light affect plant growth responses? -- How does an organism's structure help it maintain homeostasis? -- How are form and function related in the digestive system? -- How is mammalian heart structure related to function? -- How do we breathe, and why do we breathe? -- How does the immune system keep the body free of pathogens? -- What is nitrogenous waste, and how is it removed from the body? -- How do hormones regulate cell functions? -- How does the production of male and female gametes differ in humans? -- What common events occur in the early development of animals? -- How do neurons function to transmit information? -- What would happen if you modified a particular aspect of neuron function? -- How does sarcomere structure affect muscle function? -- What would happen if you modified particular aspects of muscle function? -- What factors determine climate? -- What determines behavior? -- What methods can you use to determine population density and distribution? -- What models can you use to calculate how quickly a population can grow? -- What do you need to consider when analyzing communities of organisms? -- What limits do available solar radiation and nutrients place on carrying capacities? -- What factors can affect the survival of a species or

community? The activities of this workbook focus on key ideas, principles and concepts that are basic to understanding biology. The overall organization follows that of Campbell/Reece, *Biology*, 7th edition.-p. vii. General, Organic, and Biological Chemistry *Pearson A Concise Introduction to General, Organic, and Biological Chemistry* General, Organic, and Biological Chemistry strengthens the evidenced strategy of integrating general, organic, and biological chemistry for a focused introduction to the fundamental connections between chemistry and life. The streamlined approach offers readers a clear path through the content over a single semester. The Third Edition integrates essential topics more effectively than any text on the market, covering core concepts in each discipline in just 12 comprehensive chapters. Practical connections and applications show readers how to use their understanding of chemistry in everyday life and future health professions. With an emphasis on problem solving and critical thinking, the book promotes active and attentive learning, which now include NEW! media assets, Practicing the Concepts. Featuring coauthor Todd Deal, these 3 to 5 minute videos explore key concepts in general, organic, and biological chemistry that readers traditionally find difficult. Readers gain skills and deepen their knowledge as they watch the videos and then practice what they have learned with Pause & Predict problems and a series of follow up multiple-choice questions. The Third Edition places a greater emphasis on matching what professors teach in the classroom by increasing the coverage of biochemical applications in each chapter. A new design was created to highlight the career content in order to increase relevancy. Also available as a Pearson eText or packaged with Mastering Chemistry Pearson eText is a simple-to-use, mobile-optimized, personalized reading experience that can be adopted on its own as the main course material. It lets students highlight, take notes, and review key vocabulary all in one place, even when offline. Seamlessly integrated videos and other rich media engage students and give them access to the help they need, when they need it. Educators can easily share their own notes with students so they see the connection between their eText and what they learn in class - motivating them to keep reading, and keep learning. Mastering combines trusted author content with digital tools and a flexible platform to personalize the learning experience and improve results for each student. Built for, and directly tied to the text, Mastering Chemistry enables an extension of learning, allowing students a platform to practice, learn, and apply outside of the classroom. 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Maintaining the clear and approachable writing style characteristic of author Karen Timberlake, *Basic Chemistry, Fourth Edition*, adds to its suite of problem-solving tools and techniques necessary for educational and workforce success. Engaging new features such as Key Math Skills and Core Chemistry Skills, and Math Review Modules allow students of various levels and learning styles to practice and master quantitative skills. Popular features, including "Combining Ideas" sections and end-of-chapter questions, have also been strengthened and expanded. Modern real-world applications help students connect chemical principles to events in their world, while stories involving careers illustrate the importance of chemistry in future careers. *Chemistry For Changing Times: Pearson New International Edition* *Pearson Higher Ed* Were you looking for the book with access to MasteringChemistry? This product is the book alone, and does NOT come with access to MasteringChemistry. 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The main objective in writing this text is to make the study of chemistry an engaging and a positive experience for students by relating the structure and behaviour of matter to real life. This new edition introduces more problem-solving strategies, more problem-solving guides, new features in the Sample Problems; Try It First before the Solution and Connect in the Analyse the Problem step of the solution, a new Engage feature, new conceptual and challenge problems, and new sets of combined problems. The full text downloaded to your computer With eBooks you can: search for key concepts, words and phrases make highlights and notes as you study share your notes with friends eBooks are downloaded to your computer and accessible either offline through the Bookshelf (available as a free download), available online and also via the iPad and Android apps. Upon purchase, you'll gain instant access to this eBook. Time limit The eBooks products do not have an expiry date. 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Coverage also includes: fundamentals of classical thermodynamics of phase equilibria; thermodynamic properties from volumetric data; intermolecular forces; fugacities in gas and liquid mixtures; solubilities of gases and solids in liquids; high-pressure phase equilibria; virial coefficients for quantum gases; and much more. Throughout, **Molecular Thermodynamics of Fluid-Phase Equilibria** strikes a perfect balance between empirical techniques and theory, and is replete with useful examples and experimental data. 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Tro's unprecedented two- and three-column problem-solving approach is used throughout to give students sufficient practice in this fundamental skill. A unique integration of macroscopic, molecular, and symbolic illustrations helps students to visualize the various dimensions of chemistry; Tro's engaging writing style captures student's attention with relevant applications. The Second Edition offers a wealth of new and revised problems, approximately 50 new conceptual connections, an updated art program throughout, and is available with MasteringChemistry®, the most advanced online tutorial and assessment program available. 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